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(54) Process and reactor for carrying out non-adiabatic catalytic reactions

(57) Process for carrying out non-adiabatic reactions comprising the steps of:

introducing in parallel a first stream of reactants into a first reaction space and a second stream of reactants into a second reaction space;
at reaction conditions contacting the first reactant stream with a catalyst in the first reaction space in indirect heat exchange with a heat exchanging medium and contacting the second reactant stream with a catalyst in the second reaction space in indirect heat exchange with a heat exchanging medium, and withdrawing a first and second steam reformed product gas; and
the catalyst in the first reaction space being arranged within a tubular reactor in indirect heat exchanging relationship with the heat exchanging medium by introducing the medium into tubular heat exchange space concentrically surrounding the tubular reactor with the first reaction space, the catalyst in the second reaction space being arranged on shell side of a heat exchange space in indirect heat exchanging relationship with the heat exchanging medium.

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Description

[0001] The present invention relates to a process and reactor system for carrying out non-adiabatic reactions proceeding in a process gas in presence of a catalyst exothermically or endothermically in indirect heat exchange with an appropriate heat exchange medium.

[0002] A general object of this invention is thus to provide a process for carrying out non-adiabatic reactions comprising the steps of:

introducing in parallel a first stream of reactants into a first reaction space and a second stream of reactants into a second reaction space,

at reaction conditions contacting the first reactant stream with a catalyst in the first reaction space in indirect heat exchange with a heat exchanging medium and contacting the second reactant stream with a catalyst in the second reaction space in indirect heat exchange with the heat exchanging medium, the catalyst in the first reaction space being arranged within a tubular reactor in indirect heat exchanging relationship with the heat exchanging medium by introducing the medium into tubular heat exchange space concentrically surrounding the tubular reactor with the first reaction space, the catalyst in the second reaction space being arranged on shell side of a heat exchange space in indirect heat exchanging relationship with the heat exchanging medium.

[0003] The invention is in particular useful in carrying out steam reforming reactions in a hydrocarbon feed stock by heat supplied from hot effluent gas from an autothermal steam reforming reactor and steam reformed product gas from the process.

[0004] A specific embodiment of the reaction system according to the invention is described more detailed in the following description by reference to the drawings in which Fig. 1 shows schematically a reaction system being used in the production of a gas with a high content of hydrogen and/or carbon monoxide from steam-reforming of a hydrocarbon feed stock.

[0005] Steam reforming is an endothermic chemical reaction, where hydrocarbons and steam react on a steam reforming catalyst, and if appropriate heat is supplied to the location of the reaction.

[0006] The reactor system being used in this embodiment consists of three reactors, wherein the steam reforming process is carried through. The three reactors RI, R2 and R3 are operated in parallel.

[0007] R1 is an adiabatic reactor. The reactants for the process in RI consist of hydrocarbon, steam and an oxygen rich gas being introduced into the reactor at an appropriate temperature and mixed. The oxygen and the hydrocarbon will react by combustion and result in a hot gas of residual hydrocarbon, steam and resulting in products from th combustion. Subsequently, the hot gas is passed through a bed of reforming catalyst and catalytically converted to a hot mixture of hydrogen, carbon monoxide and carbon dioxide.

[0008] R2 and R3 are two plug flow reactors. The reactants for the process in R2 and R3 are a mixture of hydrocarbon and steam, which is heated to an appropriate temperature before flowing through a bed of reforming catalyst. Walls surround and enclose the catalyst beds of R2 and R3. A hot gas is flowing outside these walls countercurrent to the reacting gases in the catalyst beds. Heat is conducted through the walls from the hot gas to the reacting gases, while the gases are converted to a hot mixture of hydrogen, carbon monoxide and carbon dioxide.

[0009] The product gases from RI, R2 and R3 are mixed and form the hot gas flowing outside the walls of R2 and R3, where they form the heat source of the reactions in R2 and R3. This gas is called the heating gas.

[0010] As a general advantage of the invention, the walls of R2 and R3 can be arranged in a layout so as to form an optimal channel for the heating gas.

[0011] The invention provides, furthermore, a reaction system being in particular useful for carrying out the above process. In general, the reaction system of this invention comprises connected in parallel a first and a second reaction compartment being adapted to hold a catalyst and to receive a reactant stream, the first compartment being in form of a reactor tube, wherein

[0012] a first heat exchange space concentric and spaced apart surrounds the first reaction compartment, and the second reaction compartment surrounds a second heat exchange space.

[0013] Reactor R2 contains the catalyst inside tubes. Reactor R3 holds the catalyst outside the tubes. The combined reactor R2 and R3 comprises a number of double-tubes, where the inner tubes are catalyst filled (R2) and the double-tubes are in addition arranged in a pattern allowing the volume between the double-tubes to be filled with catalyst as well, i.e. reactor R3. The sensible heat from the combined product gas from the reactors R1, R2 and R3 is cycled back to the reactors R2 and R3. The product gas is flowing in annular channels provided by the double-tubes, countercurrently to the flow in the reactors R2 and R3. Heat is supplied to reactor R2 via the inner wall of the double pipes and the reactor R3 is supplied with heat from the outer wall of the double-tubes.

[0014] The advantage of the combined reactor as shown in Fig. 2 is that the heat exchange channels are utilised in an optimal manner, i.e. both the inner wall and the outer wall are utilised as exchange heat surfaces thus making optimal use of expensive material. This also leads to a very compact design of equipment compared to other types of heat exchange reformers and at the same time provides low pressure drop.

[0015] On cooling the product gas, a certain risk of metal dusting corrosion exists. A further advantage of

the combined reactor design is restricted risk of metal dusting to a limited surface.

[0016] The double tube dimensions are typically: Inner tube OD 50 to 140 mm and outer tube OD 80 to 170 mm. The layout can be but need not be arranged in such a way that the heat exchange/ area/catalyst volume ratio is equal for the inner tubes and the outer tubes.

Claims

1. Process for carrying out non-adiabatic reactions comprising the steps of:

   introducing in parallel a first stream of reactants into a first reactions space and a second stream of reactants into a second reaction space;
   at reaction conditions contacting the first reactant stream with a catalyst in the first reaction space in indirect heat exchange with a heat exchanging medium and contacting the second reactant stream with a catalyst in the second reaction space in indirect heat exchange with a heat exchanging medium, and withdrawing a first and second steam reformed product gas; and
   the catalyst in the first reaction space being arranged within a tubular reactor in indirect heat exchanging relationship with the heat exchanging medium by introducing the medium into tubular heat exchange space concentrically surrounding the tubular reactor with the first reaction space, the catalyst in the second reaction space being arranged on shell side of a heat exchange space in indirect heat exchanging relationship with the heat exchanging medium.

2. Process of claim 1, wherein the non-adiabatic ray action is endothermic steam reforming of a hydrocarbon feedstock.

3. Process of claim 1, wherein the heat-exchanging medium comprises an effluent stream from autothermal steam reforming of a hydrocarbon feed stock and/or the product gas.

4. Reaction system for carrying out non-adiabatic catalytic reactions, comprising connected in parallel a first and second reaction compartment being adapted to hold a catalyst and to receive a reactant stream, the first compartment being in form of a reactor tube, wherein
   a first heat exchange space concentric and spaced apart surrounds the first reaction compartment, and the second reaction compartment surrounds a second heat exchange space.

5. Reaction system of claim 4, wherein the first and second reaction compartment are arranged within a common shell.

6. Reaction system of claim 4, wherein the first and second heat exchange space are formed by a common passageway.

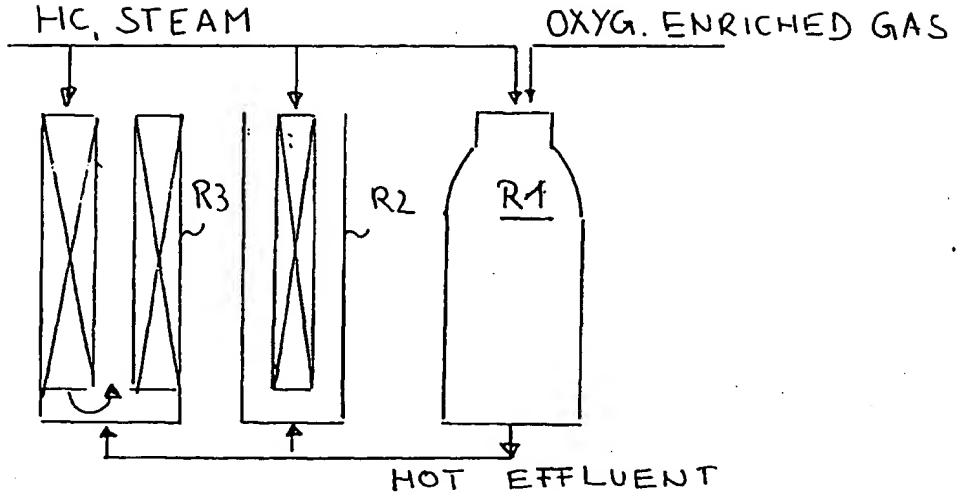
HC, STEAM
OXYG. ENRICHED GAS
R3
R2
R1
HOT EFFLUENT

Fig. 1

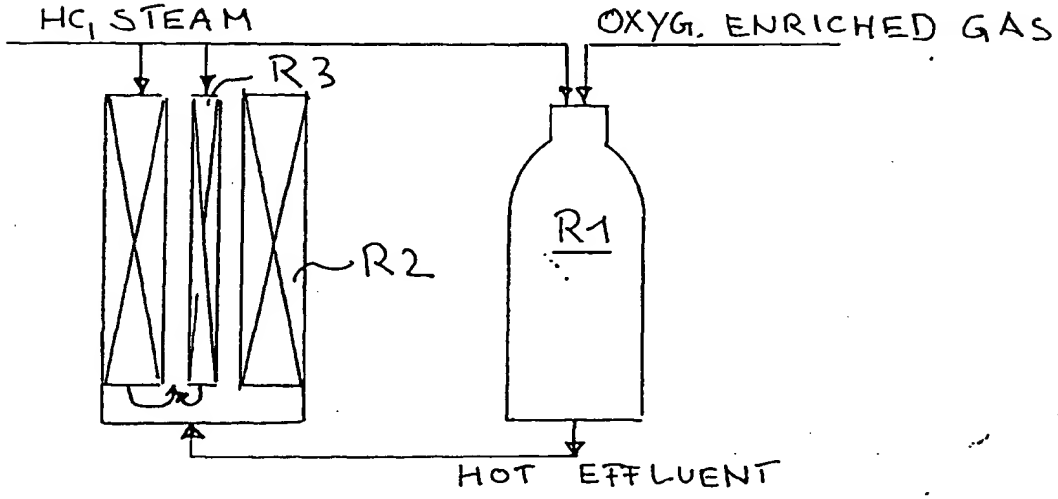
HC, STEAM
OXYG. ENRICHED GAS
R3
R2
R1
HOT EFFLUENT

Fig. 2

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EUROPEAN SEARCH REPORT

Application Number
EP 00 12 5077

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The present search report has been drawn up for all claims

Place of search	Date of completion of the search	Examiner
THE HAGUE	7 February 2003	Van der Poel, W

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X : particularly relevant if taken alone
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